

Dimethyl(methylthio)sulfonium Tetrafluoroborate: A Reagent for Disulfide Bond Formation in Peptides

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Abstract: Dimethyl(methylthio)sulfonium tetrafluoroborate (1) has been used to deprotect a series of cysteine derivatives (BocCys-Acm, MeOBzl, Trt) with concurrent disulfide bond formation. This reagent has also been applied to the deprotection and disulfide bond formation in two peptides: a sarafotoxin A fragment and an analog of calcitonin where each peptide contains two S-protected cysteine derivatives: [Cys(Acm)]

Disulfide linkages are a common means by which the tertiary structure of small proteins are stabilized.¹ Likewise disulfide bonds between two cysteine residues have been used to decrease the conformational flexibility of peptides in solution.² Although disulfide bonds have been widely used in peptide and protein chemistry, there exist only a limited number of methods available for their synthesis.³ Many of these reagents are less than ideal due to undesired functionalization of amino acid sidechains by the reagents. In this paper we report the use of dimethyl(methylthio)sulfonium tetrafluoroborate⁴ (1) as a selective and mild reagent used for removing cysteine S-protecting groups with concomitant disulfide bond formation in amino acid derivatives and peptides.

The ability of this reagent to remove a variety of cysteine protecting groups was initially examined using four Boc-Cys derivatives with different S-protecting groups (Acm, Bzl, MeOBzl, Trt). The product derived from the reaction of 1 with 2a, 2c and 2d was the mixed disulfide 3 (Figure 1). With the benzyl derivative 2b, only starting material was obtained after 72 hrs of reaction.

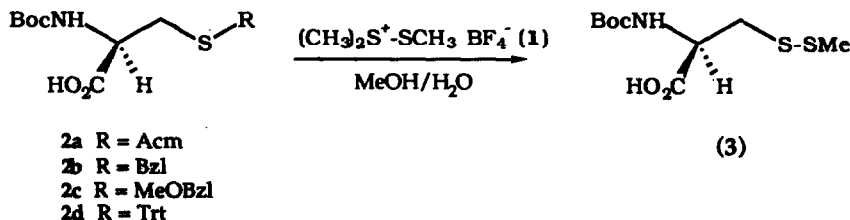


Figure 1. Deprotection and mixed disulfide formation of Cys derivatives with (1).

In a standard reaction, reagent 1 (10 eq) was added to a solution of an amino acid (2a-d) in MeOH/H₂O (1:1, 10 mM). After 24 hr the reaction was purified directly by reverse phase HPLC. With cysteine derivatives 2a(Acm), 2c(MeOBzl), and 2d(Trt) all starting material was consumed within 24 hr with the exclusive appearance of the mixed disulfide 3 in approximately 50% yield after chromatography.⁵ Derivative 2b was recovered unchanged from the reaction mixture.

Reagent 1 was also examined for its ability to deprotect two S-protected cysteine residues within a single peptide sequence with concomitant intramolecular disulfide bond formation (Figure 2). An analog of calcitonin (4) was used in this study with the Ac_m S-protecting group due to the facile reaction of Ac_m with 1 in the above model studies.⁶

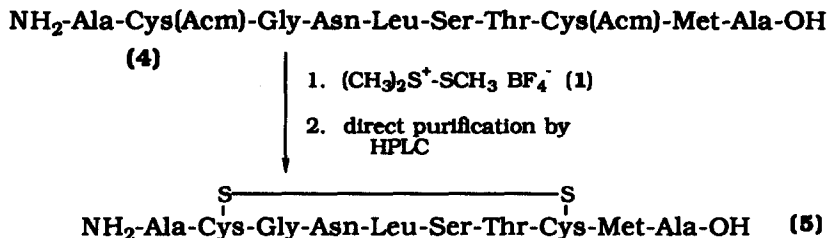


Figure 2. Deprotection and intramolecular disulfide bond formation in calcitonin analogs.

In a typical reaction a solution of peptide 4 (mixed aqueous/organic solvent systems, 0.1 or 1 mM solutions) was treated with 1 (10 eq) until complete as monitored by analytical HPLC. The reactions were purified directly by preparative HPLC. All reported yields of 5 are isolated yields after preparative HPLC (Table 1).^{7,8}

Peptide	Reagent	Conc. ^a	Solvent ^b	Time	Product	Yield ^c
4	1	1mM	MeOH/H ₂ O	24 hr	5	28%
4	1	0.1mM	MeOH/H ₂ O	24 hr	5	37%
4	1	0.1mM	TFE/H ₂ O	24 hr	5	46%
6	1	1 mM	MeOH/H ₂ O	24 hr	7	59%
6	1	1 mM	MeOH/H ₂ O	24 hr	8	35%
6	1	1 mM	H ₂ O	24 hr	8	43%
6	I ₂	1 mM	MeOH/H ₂ O	24 hr	8	28%

^aFinal concentration of peptides.
^b1:1 mixture of organic solvent in H₂O.
^cIsolated yields of peptides after preparative HPLC.

A longer peptide sequence corresponding to a fragment of sarafotoxin A (6) was examined for its reactivity with reagent 1.⁶ In this case the peptide containing the mixed disulfides with methyl mercaptan (7) was obtained upon treatment with 1. Peptide 7 was converted into the cyclic analog (8) by treatment with a Reduce-Imm™ reducing column (Pierce), followed by oxidation with K₃Fe(CN)₆ (Figure 3).^{7,8}

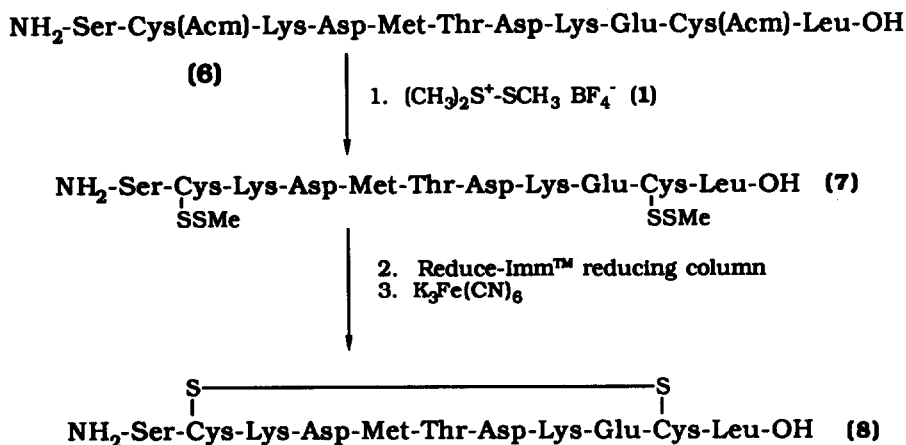


Figure 3. Deprotection and intramolecular disulfide bond formation in a fragment of sarafotoxin A.

In a typical reaction a solution of peptide 6 (aqueous or mixed aqueous/organic solvent system, 1mM solutions) and was treated with 1 (10 eq). The mixture was stirred at room temperature for 24 hr and purified by reverse phase HPLC to yield 7 (Table 1). Peptide 7 was dissolved in a 10 mM phosphate buffer (pH 8) and run through a Reduce-ImmTM column. The fractions which reacted positively with Ellman's reagent⁹ were pooled and oxidized with K₃Fe(CN)₆ to yield the cyclic peptide 8. Alternatively, the crude solution of 7 was concentrated to dryness *in vacuo*, the residue dissolved in a 10 mM phosphate buffer (pH 8) and reduced directly with the Reduce-ImmTM column, followed by oxidation with K₃Fe(CN)₆.

We have identified two different types of reaction products which are obtained from the treatment of amino acids and peptides with dimethyl(methylthio)sulfonium tetrafluoroborate (1). In the case of amino acids containing one S-protecting group only the mixed disulfide with methyl mercaptan is obtained. When peptides containing two S-protecting groups are treated with 1, the cyclic peptide was obtained directly in a case where five amino acids residues intervened between the cysteine residues. In peptides with seven and eight¹⁰ residues between the cysteines only the mixed disulfides with methyl mercaptan were obtained upon treatment with 1, perhaps due to a lack of intramolecular disulfide exchange in the larger ring systems. In these cases the cyclic peptides were easily obtained from a DTT reduction/K₃Fe(CN)₆ oxidation sequence.

From an investigation of the reactivity of 1 with different S-protecting groups, we have determined that protecting groups which yield stable carbocations - such as Acm, MeOBzl, and Trt - react to give the methyl mercaptan mixed disulfide, whereas no reaction was detected with the Bzl protecting group. Reagent 1 is more versatile than other reagents used for forming disulfides in peptides, such as I₂, in that it is capable of removing the MeOBzl protecting group, can be used in a wider range of solvents (including H₂O), and gives a higher yield of deprotected peptides. In conclusion, 1 is a mild reagent for the removal of a wide range of S-protecting groups with concomitant disulfide bond formation, and as such it is a valuable addition to the methods available for disulfide bond formation in peptides.

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Abbreviations: Acn = acetamidomethyl, AcOH = acetic acid, Bzl = benzyl, DTT = dithiothreitol, MeOBzl = 4-methoxybenzyl, Trt = trityl, TFE = trifluoroethanol

References and Notes

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5. All reaction mixtures were purified by HPLC using a Vydac-C₈ reverse phase column (2.2 x 25 cm) with mobile phase A (100% CH₃CN/0.1% TFA) and mobile phase B (100% H₂O/0.1% TFA) with a linear gradient (5-95% A) of 60 min. All amino acid derivatives were characterized by ¹H NMR and mass spectroscopy (FAB, glycerol matrix).
6. Peptides were synthesized using solid phase methods on the *p*-alkoxybenzyl alcohol resin (Wang, S. S., *J. Am. Chem. Soc.* **1973**, *95*, 1328) with the conventional fluorenylmethyloxycarbonyl-based strategy (Atherton, E.; Sheppard, R. C., in *The Peptides*; Vol. 9; Gross, E.; Meienhofer, J. Eds.; Academic Press: New York, 1987; pp. 1-38).
7. **HPLC conditions:** All reactions were monitored and purified by HPLC using Vydac-C₈ reverse phase columns (analytical - 1 x 25 cm, preparative - 2.2 x 25 cm) with mobile phase A (100% CH₃CN/0.1% TFA) and mobile phase B (100% H₂O/0.1% TFA) with linear gradients (5-60% A) of 30 min (analytical) or 60 min (preparative).
8. All new peptides were identified by mass spectroscopy (FAB, glycerol matrix) and amino acid analysis.
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10. Reagent **1** was also tested on a fragment of epidermal growth factor containing two Cys(Acn) derivatives (data not shown).

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